

1- ہر سوال کے سامنے چار دائرے دئے گئے ہیں، صرف صحیح جواب والا دائرہ بھریں۔

2- دائروں کو شیڈ (بھرنے) کے لئے نیلے یا کالے رنگ کا مارکر استعمال کریں۔

3- جواب میں ایک سے زائد دائرے بھرنے سے جواب غلط تصور ہوگا۔

Time Allowed: 20 Minutes

SECTION – A

Marks : 18

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|----|--|---|--|--|---|
| 1 | The largest no of molecules is present in | <input type="radio"/> 22 g of CO ₂ | <input type="radio"/> 49 g of H ₂ SO ₄ | <input type="radio"/> 18 g of H ₂ O | <input type="radio"/> 90 g of C ₆ H ₁₂ O ₆ |
| 2 | The mass of one molecule of H ₂ O is | <input type="radio"/> 18 g | <input type="radio"/> 0.18 g | <input type="radio"/> $\frac{18}{6.02 \times 10^{23}}$ g | <input type="radio"/> $\frac{6.02 \times 10^{23}}{18}$ g |
| 3 | How many pairs of electrons surround the central atom of "Sn" in the angular SnCl ₂ molecule? | <input type="radio"/> 1 | <input type="radio"/> 2 | <input type="radio"/> 3 | <input type="radio"/> 4 |
| 4 | The energy carried by a photon is given by: | <input type="radio"/> E = hν | <input type="radio"/> E = $\frac{hc}{\lambda}$ | <input type="radio"/> Both A and B | <input type="radio"/> None of these |
| 5 | The e/m ration for canal rays particles is than that of cathode ray particles. | <input type="radio"/> Smaller | <input type="radio"/> Greater | <input type="radio"/> Equal | <input type="radio"/> May be greater or smaller |
| 6 | Which bonds are responsible for hair to be straight or curly? | <input type="radio"/> Hydrogen bonding | <input type="radio"/> Disulphide bridge | <input type="radio"/> Glycosidic linkage | <input type="radio"/> Peptide linkage |
| 7 | Which of the given gases can be liquefied above its critical temperature? | <input type="radio"/> O ₂ | <input type="radio"/> N ₂ | <input type="radio"/> H ₂ | <input type="radio"/> None of these |
| 8 | If the mass of electron is doubled, the Rydberg constant: | <input type="radio"/> Becomes half | <input type="radio"/> Becomes double | <input type="radio"/> Remains unchanged | <input type="radio"/> Becomes one fourth |
| 9 | Hydrolysis of a salt may result in solution. | <input type="radio"/> Acidic | <input type="radio"/> Basic | <input type="radio"/> Neutral | <input type="radio"/> Any of the given |
| 10 | The value of p _{kw} with increase in T. | <input type="radio"/> Decreases | <input type="radio"/> Increases | <input type="radio"/> Remains constant | <input type="radio"/> Both A & B |
| 11 | A crystal can have centre / centres of symmetry at maximum. | <input type="radio"/> 1 | <input type="radio"/> 2 | <input type="radio"/> 3 | <input type="radio"/> 4 |
| 12 | The rate of evaporation of a liquid increases by | <input type="radio"/> Increasing surface area | <input type="radio"/> Decreasing intermolecular forces | <input type="radio"/> Increasing temperature | <input type="radio"/> All these |
| 13 | Rate = K[A] ^m [B] ⁿ . This mathematical expression is called | <input type="radio"/> Rate law | <input type="radio"/> Rate expression | <input type="radio"/> Rate equation | <input type="radio"/> All these |
| 14 | Daniell cell is not cell. | <input type="radio"/> An electrochemical | <input type="radio"/> A Galvanic | <input type="radio"/> An electrolytic | <input type="radio"/> A voltaic |
| 15 | Which of the given is not a colligative property of a solution? | <input type="radio"/> Lowering of vapour pressure | <input type="radio"/> Elevation in B.P° | <input type="radio"/> Osmotic pressure | <input type="radio"/> Mole fraction |
| 16 | Which of the given is the most strongest? | <input type="radio"/> Hydrogen bonding | <input type="radio"/> Dipole-dipole interaction | <input type="radio"/> London dispersion forces | <input type="radio"/> Intramolecular forces |
| 17 | Pepsin is found in | <input type="radio"/> Salvia | <input type="radio"/> Gastric Juice | <input type="radio"/> Bile extract | <input type="radio"/> Pancreatic juice |
| 18 | For an ideal gas the compressibility factor has a value of | <input type="radio"/> Zero | <input type="radio"/> Unity | <input type="radio"/> Two | <input type="radio"/> Infinity |